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Obtaining hydrophobic coatings from AR+HMDSO using radiofrequency discharge at atmospheric pressure

Abstract. Hydrophobic coatings have attracted much attention due to their wide applications in various industries including electronics, textiles, and automotive. This study deals with the process of creating nanoscale coatings on a substrate achieved by plasma polymerization using plasma flow (jet) using radiofrequency discharge together with the precursor $\text{C}_6\text{H}_{18}\text{OSi}_2$ and the carrier gas Ar. In this work, we investigate the production of hydrophobic coatings using radiofrequency (RF) discharge at atmospheric pressure using a mixture of argon (Ar) and hexamethyldisiloxane (HMDSO). RF discharge is a versatile and efficient method of plasma generation that allows the deposition of thin films with defined properties. Here we investigate the influence of process parameters such as gas flow rate, discharge power, and substrate temperature on the morphology, chemistry, and hydrophobicity of the deposited coatings. The formation of these coatings was carefully studied under atmospheric pressure conditions, varying the number of cycles of experiments while maintaining optimal plasma parameters. The properties and elemental composition of the coatings were thoroughly studied using scanning electron microscopy (SEM) and energy dispersion spectroscopy (EDS). In addition, the obtained coatings were found to possess hydrophobic properties. The hydrophobicity of these coatings was evaluated by measuring the contact angle with a goniometer with respect to cycles of experiments and long-term durability. This study contributes to a better understanding of the synthesis, structure, and hydrophobic characteristics of nanoscale coatings, opening promising perspectives for various applications. The results show that the hydrophobicity of the coatings can be optimized by tuning the process parameters, resulting in coatings with water contact angles greater than 160 degrees. Additionally, the durability and stability of the hydrophobic coatings are evaluated under different environmental conditions, allowing an assessment of their potential for practical applications.

Key words: plasma polymerization, atmospheric pressure, superhydrophobic coatings.

Introduction

Hydrophobic coatings, which repel water due to their low surface energy, find wide applications in various industries ranging from electronics to textiles [1], automotive [2], and medical devices [3]. Among the methods for producing hydrophobic coatings, the radiofrequency (RF) discharge at atmospheric pressure offers a promising avenue [4]. This technique involves the deposition of hydrophobic coatings from a precursor gas mixture of argon (Ar) and hexamethyldisiloxane (HMDSO) under the influence of RF discharge. The use of RF discharge at atmospheric pressure presents several

advantages over conventional methods such as chemical vapor deposition (CVD) [5] and plasma-enhanced chemical vapor deposition (PECVD) [6]. Firstly, operating at atmospheric pressure eliminates the need for vacuum systems, simplifying the setup and reducing production costs. Secondly, the RF discharge process allows for precise control over the deposition parameters, resulting in coatings with tailored properties [7,8].

Certain techniques, such as acquiring coatings through chemical processes and under atmospheric pressure, offer superior cost efficiency and enhanced efficacy for large-scale manufacturing. Through the utilization of chemical procedures, it is possible

to procure hydrophobic and superhydrophobic coatings characterized by exceptional stability [9]. Nonetheless, the chemical approach for obtaining superhydrophobic coatings necessitates the use of hazardous liquids and diverse solvents, presenting a notable drawback that can endanger human health and the environment. Moreover, the aforementioned techniques entail multiple stages for achieving superhydrophobic coatings, rendering the process of uniformly dispersing coatings intricate and time-intensive. Hence, the imperative of employing economical and eco-friendly methods across expansive surfaces emerges as a pressing concern.

Hydrophobic coatings obtained from Ar+HMDSO using RF discharge exhibit excellent water-repellent properties, making them suitable for applications requiring protection against moisture [10], corrosion [11], and fouling [12]. These coatings demonstrate high durability [13], thermal stability [14], and chemical resistance [15], making them ideal for outdoor and harsh environment applications [16,17].

We delve into the mechanisms underlying the formation of hydrophobic coatings through RF discharge at atmospheric pressure using Ar+HMDSO precursor gas mixture. We explore the influence of process parameters such as gas composition, discharge power, and substrate temperature on the properties of the resulting coatings. Additionally, we highlight the potential applications and future prospects of these hydrophobic coatings in various industries.

The results obtained may be of significant practical importance for the development of new materials with improved characteristics and expansion of their application area.

Experimental section

The scheme of an experimental installation aimed at studying the coating obtained using a plasma jet based on RF discharge at atmospheric pressure is shown in Figure 1. The plasma flux (jet) was generated using an RF generator between a copper wire wrapped in quartz glass and a grounded copper plate. Glass tube length: 100mm, inner diameter: 3mm and outer diameter: 10mm. Seren-R 301 with a high frequency of 13.56 MHz was used as a power source. Inert gas Ar and precursors HMDSO were used as the main gas flow. HMDSO is in liquid form at room temperature and therefore has a small volume poured into a barboter. The flow of the prequels is controlled by the mass flow controller. The resulting coating samples settle on a glass substrate measuring

2x2 centimetre. Before starting the experiment, glass samples used as substrates are cleaned in an ultrasonic bath for 10 minutes. First, the concentration of gas mixtures Ar and HMDSO is sent to the quartz glass tube, and then, when a high-frequency voltage is applied to the electrode located outside the tube, atmospheric pressure plasma ignition occurs. As a result of coagulation of atoms in a plasma medium, the process of chemical reactions takes place, at a certain moment of time the coating grows to the nanoscale [10]. The experiment was carried out for the following parameters of the plasma: power magnitude 100-200 Watts, HMDSO/Ar fraction (Ar fraction 98%, HMDSO fraction 2%, respectively)

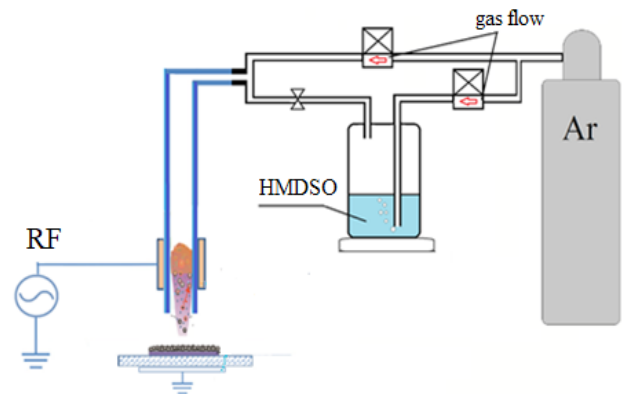


Figure 1 – Scheme of experimental setup

Main provision

In this study, a nanoscale coating was applied to the glass surface using RF discharge at atmospheric pressure. The coating exhibits hydrophobic properties. The hydrophobicity was tested under various experiment cycles while maintaining constant power, as well as assessing its durability over time.

Results and discussion

The following analyses were carried out to determine the properties of the coating formed as a result of the reaction of precursors with plasma flow on the surface of a glass sample.

- Morphological properties

The morphological properties of the synthesized samples were studied using a Quanta 3D 200i scanning electron microscope (SEM). The results of the analysis showed that the obtained particles have a variety of shapes, including spherical. The surface of the particles is heterogeneous and their sizes vary from 20 nm to 0.1 μm (Fig. 2).

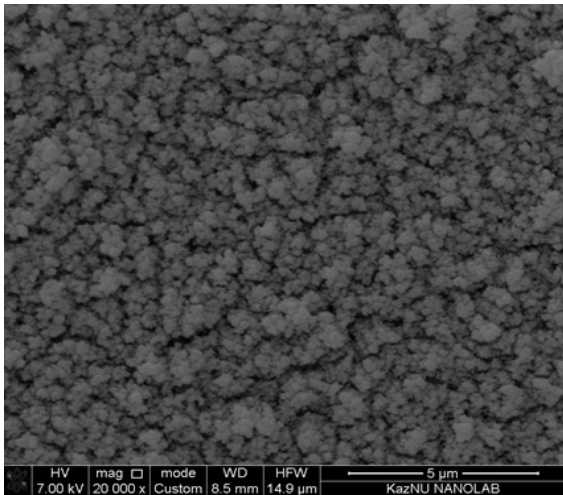


Figure 2 – SEM-micrographs of synthesized superhydrophobic particles formed on the glass surface

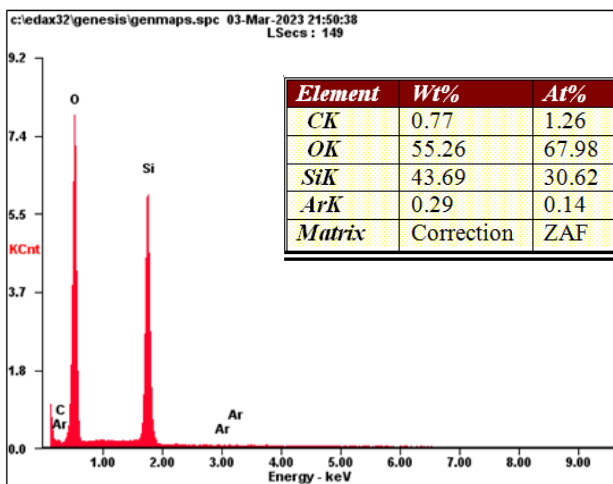


Figure 3 – Chemical composition of obtained coatings

- Chemical composition

EDS spectroscopy (Electron Dispersive Spectroscopy) was used to determine the chemical composition of the synthesized coatings. The results of the analysis showed that the obtained coatings consist of the precursors component elements (Fig. 3)

- Hydrophobic properties

The morphological analysis of the obtained sample showed that the surface presents a water repellent property, i.e. hydrophobic character. To confirm this, the angle of contact with water was evaluated. Figure 4 shows a graph showing the relationship between the contact angle and the number of cycles of the experiment.

When analyzing the graph, it becomes evident that the contact angle undergoes minimal changes when the number of cycles is changed (Fig. 5).

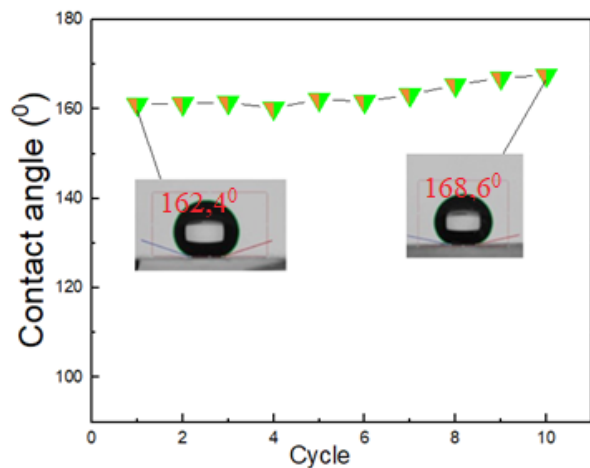


Figure 4 – Hydrophobic properties of the obtained samples proving through contact angle from the cycle of experiment

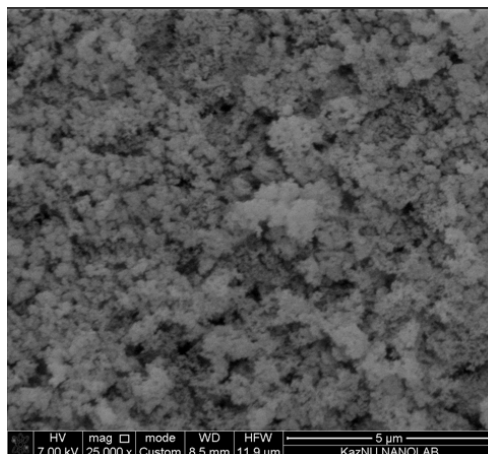
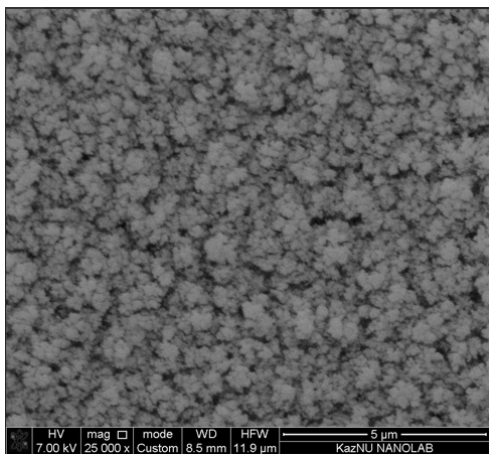


Figure 5 – Change of coating surface with increasing cycle of the experiment

This indicates that the experimental cycles have a negligible effect on the contact angle of the film because the process reached an equilibrium state relatively quickly, where further cycles have less and less effect on the surface properties. Also for further characterization, the contact angle with a long time course was measured (Figure 6). A trend is shown that the contact angle decreases over time which indicates a loss of superhydrophobicity i.e. the surface becomes less water repellent. The samples were stored under room conditions. However, when exposed to environmental factors such as UV radiation, humidity or chemical contaminants, the surface may undergo degradation. As a result of degradation, the surface structure may change, i.e., the chemical composition of the surface changes resulting in an uneven surface that will reduce its ability to effectively repel water.

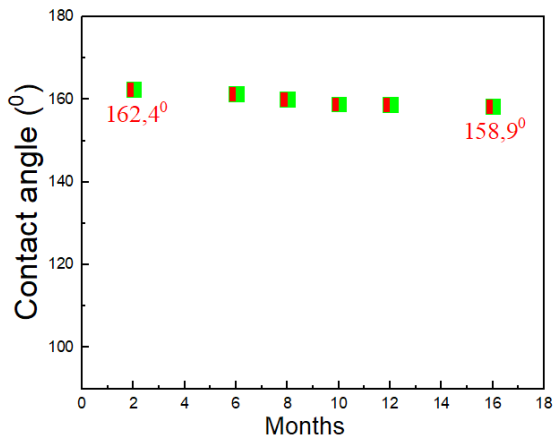


Figure 6 – Hydrophobic properties of the obtained samples proving through contact angle from the over time (0-18 months)

Conclusion

In conclusion, this study successfully demonstrated the formation and characterization of nanoscale coatings on a substrate through plasma polymerization, utilizing the precursor C₆H₁₈OSi₂ and Ar carrier gas under radiofrequency discharge. The investigation focused on optimizing plasma parameters and assessing the coatings' properties under atmospheric pressure conditions, with particular attention to the number of experiment cycles. Through thorough analysis using scanning electron microscopy (SEM) and energy dispersive spectroscopy (EDS), we gained insights into the structural morphology and elemental composition of the coatings. Furthermore, the observed hydrophobicity of the coatings suggests their potential utility in applications requiring water repellency.

The results presented herein contribute to advancing our understanding of nanoscale coating synthesis and performance, with implications for various fields such as surface engineering, materials science, and functional coatings. Future research may delve deeper into optimizing coating parameters and exploring additional characterization techniques to further enhance our understanding and potential applications of these coatings.

Acknowledgements

All authors are grateful to the Committee of Science of the Ministry of Science and Higher Education of the Republic of Kazakhstan (Grant No. AP13067865 and Program No. BR21882187)

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Received 29 April 2024

Accepted 03 June 2024